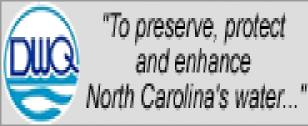
GEOCHEMICAL PROCESSES AFFECTING ARSENIC IN PRIVATE WELLS OF THE NORTH CAROLINA PIEDMONT

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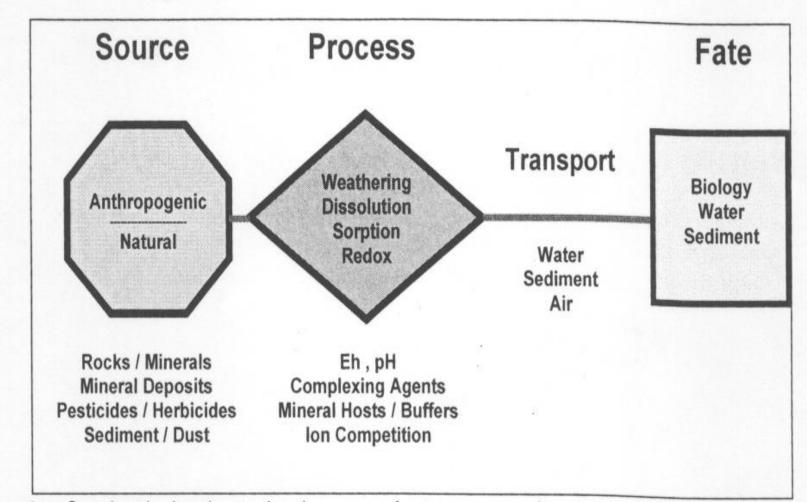


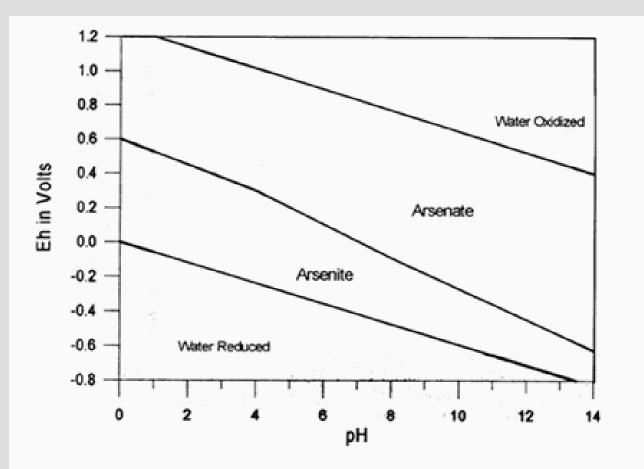
Figure 1. Geochemical pathways involve source factors, process factors that act on sources, and transport factors that collectively influence chemical concentrations in sediment, water, and biological receptors. (Robinson and Ayotte, 2007)

Arsenic:

Abundance & Speciation

(summarized from Smedley and Kinniburg, 2002)

	Baseline Conc. (ug/L)	Arsenic Species
Rain water (rural areas)	<0.03	-
Sea water - oxic	1.5	As(V) dominant
Sea water - anoxic	-	As(III) dominant
Estuarine water	0.7 - 2.5	As(V) dominant
Estuarine water - anthropogenic		As(III) dominant
Stream water - natural	0.83	As(V) dominant
Stream water - anthropogenic	-	As(III) dominant
Groundwater	< 10	variable
Groundwater - strongly reducing		As(III) dominant

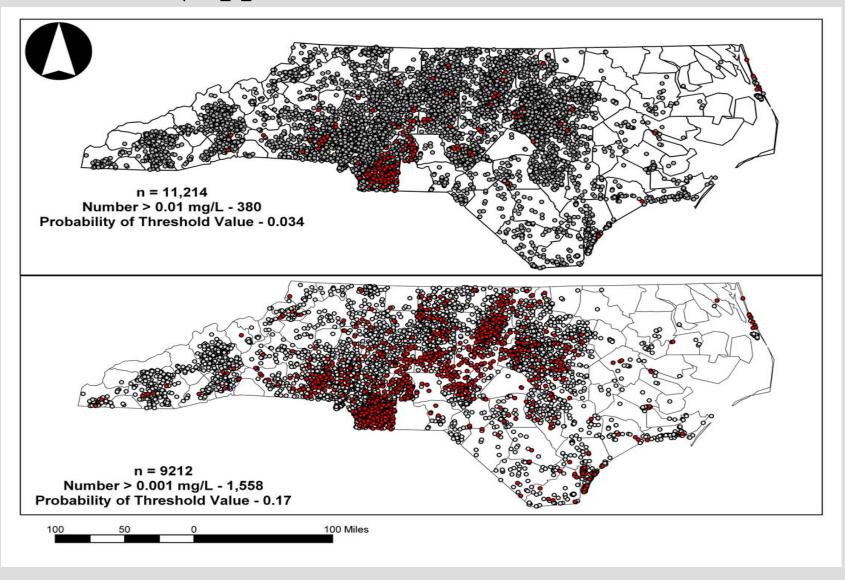


Stability fields for arsenate and arsenite

Arsenic adsorbtion to iron oxyhydroxides is strongly influenced by pH, redox potential, and presence of competing anions.

A number of factors are involved in the adsorption-desorbtion reactions. These include changes in redox potential and microbial-mediated reductive dissolution of iron hydroxides.

Previous Work in NC Piedmont (Pippin and others 2002 – 2007)



Hypothetical Groundwater Transport Model for Piedmont Aquifer System

Infiltration of O_2 rich water to groundwater surface. Dissolution of host materials and precipitation of dissolved ions.

Migration along the regolith flow path maintains an oxidative redox potential. Again dissolution and precipitation.

Migration from oxic to reducing environment. Dissolution slows. Desorption of As from oxyhydroxide become more significant.

Host Rock

0

Movement from the reducing conditions in the bedrock aquifer to greatly increased oxidative state in the discharge streams initiates the precipitation of dissolved constituents.

Migration along the bedrock flow path results in accumulation of dissolved ions.

Highly Oxidative

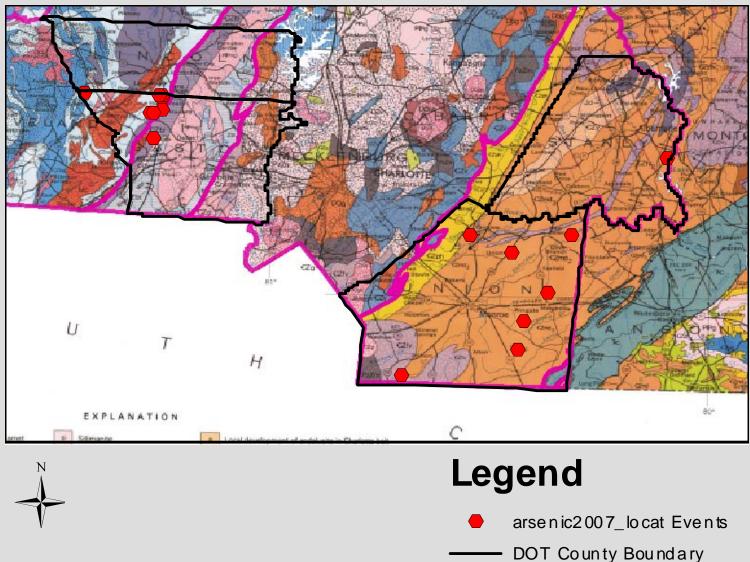
Moderately Oxidative

Mixing Zone Between Oxic and Reducing Waters

Reducing

Precipitation of Dissolved lons

STUDY AREA: GEOLOGIC MAP & SAMPLE LOCATIONS



METHODS

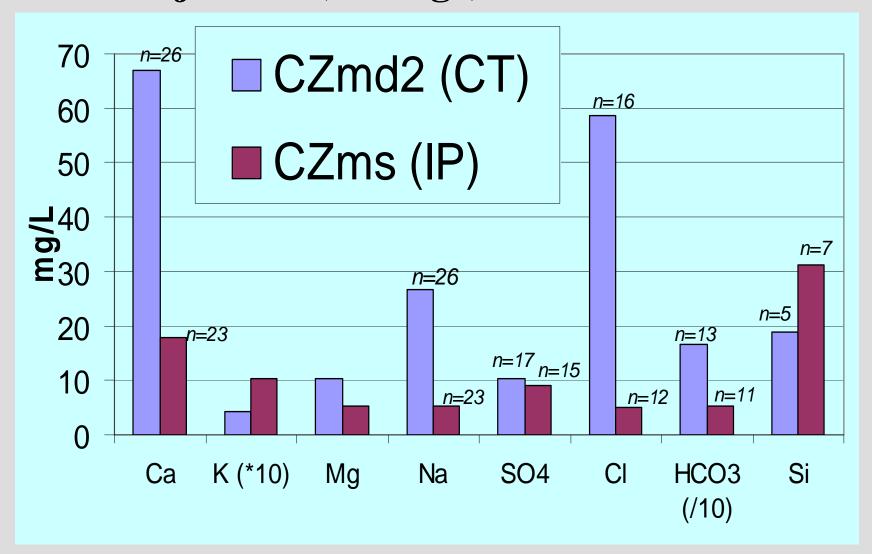
- Sample Collection
 - 3 to 5 samples from 16 locations
- Field measurements
 - pH, ORP, DO, Temp, Specific conductance
- Laboratory measurements
 - Al, Ca, Mg, Fe, K, Na
 - As, Ba, Cd, Cr, Cu, Mn, Pb, Se, Zn; Cl⁻, SO4⁻, HCO3⁻
 - Arsenic analyzed by EPA method 200.8.
 - Detection limit for As = 5 ug/L

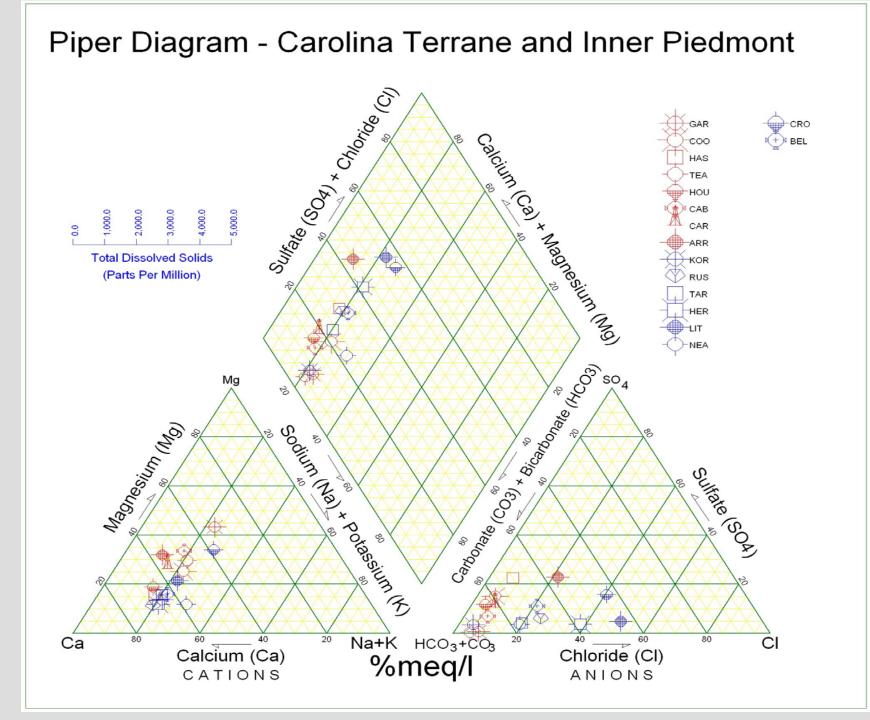
Field Speciation of Arsenic



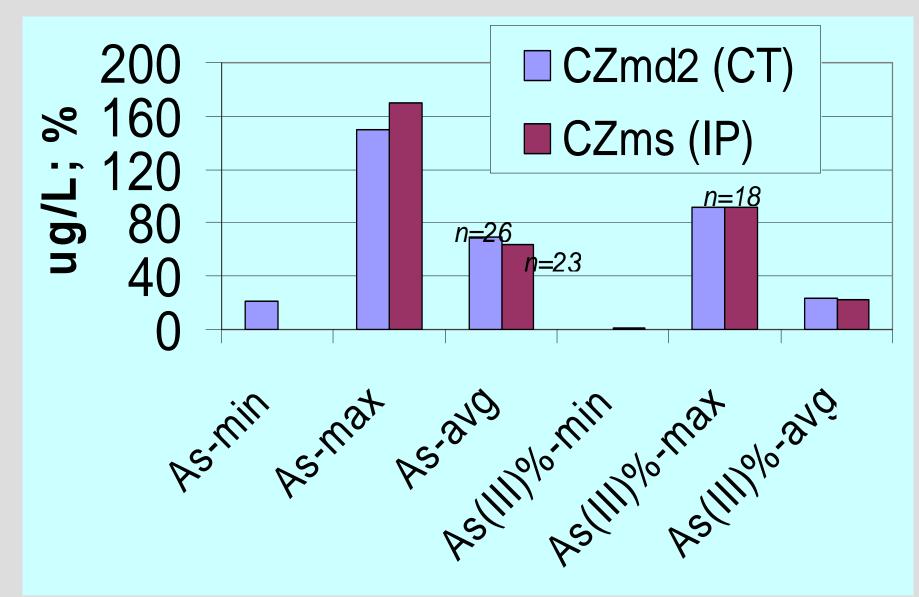
The adsorbent in the cartridge removes arsenate
 [As(V)] but does not adsorb arsenite [As(III)]

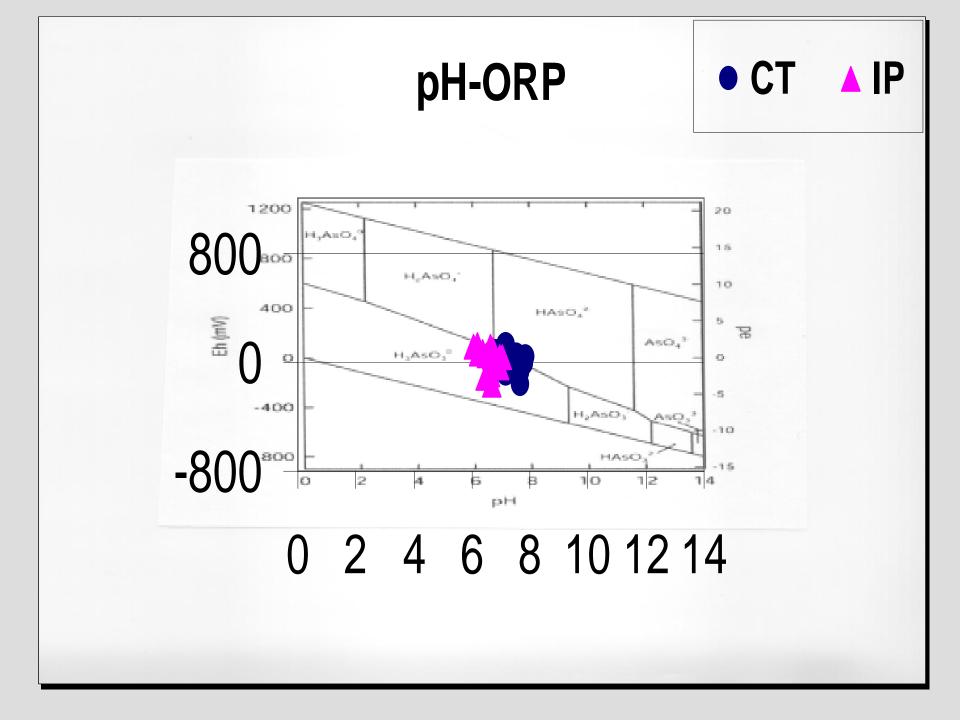
RESULTS Major Ion (average) Concentrations



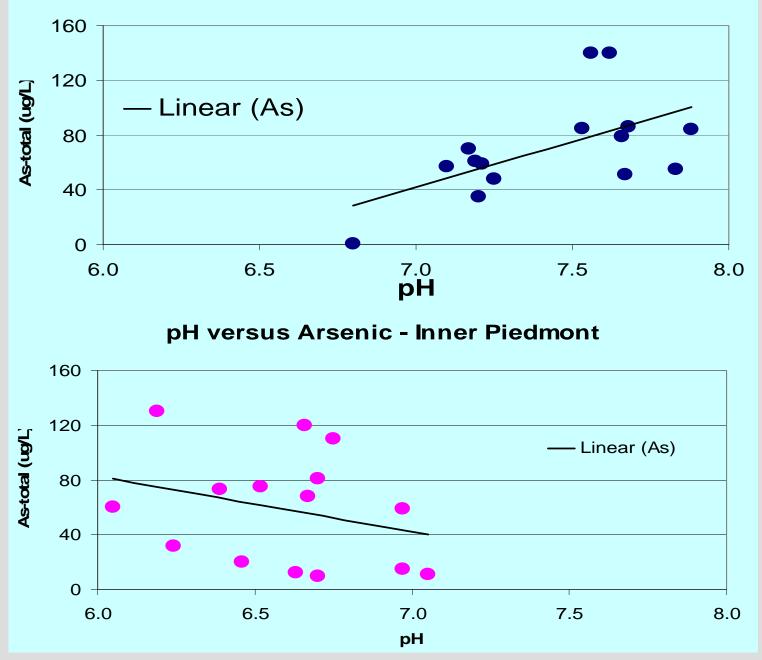


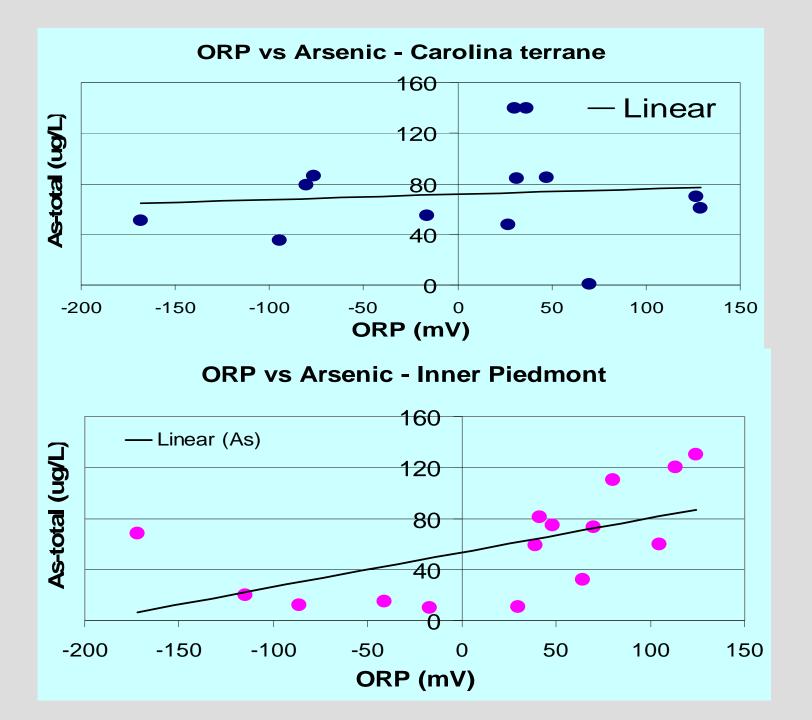
As concentration and speciation

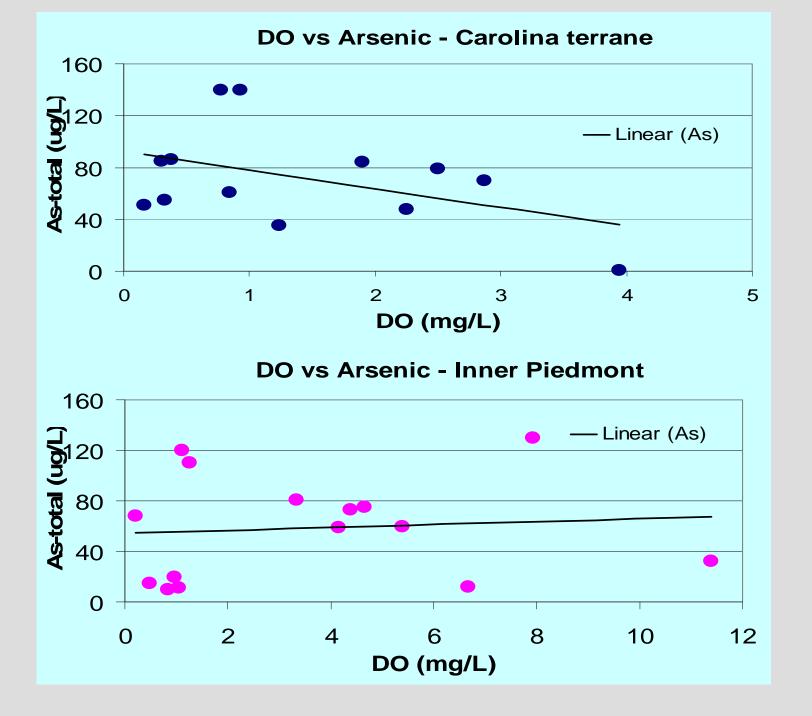




pH versus Arsenic - Carolina terrane

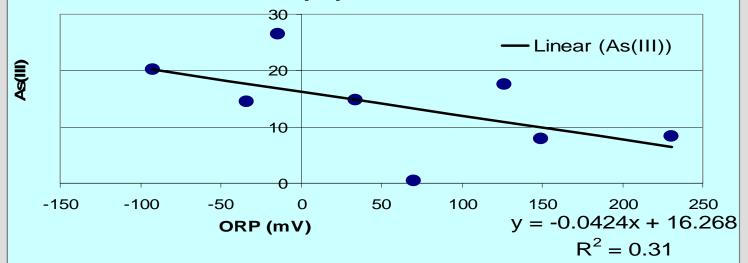




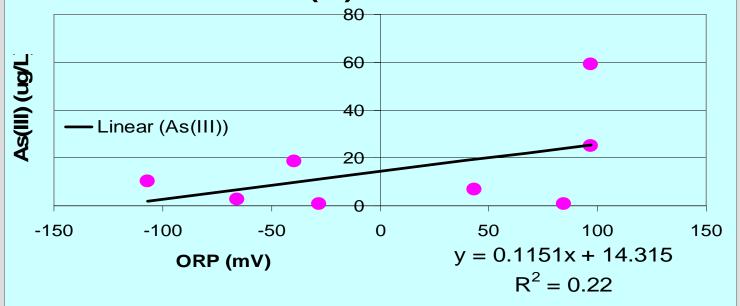


Linear Regressions

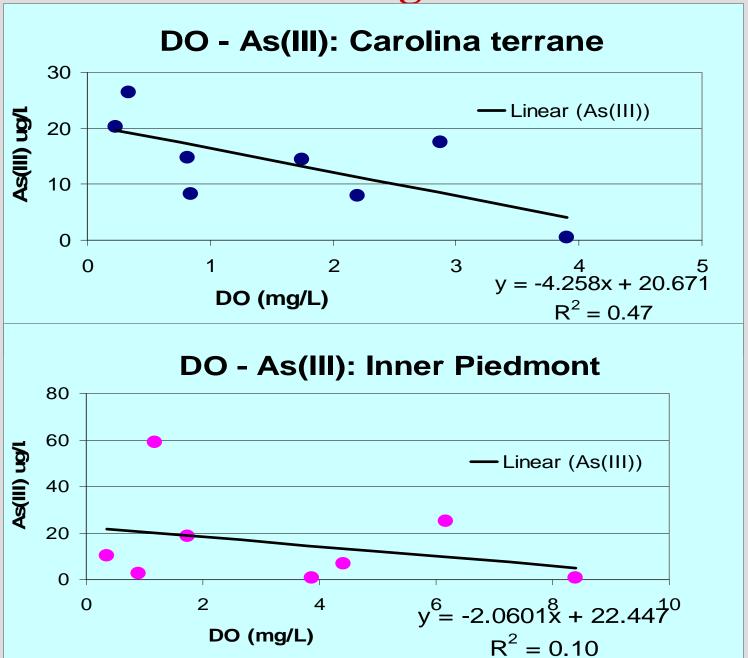
ORP - As(III): Carolina terrane



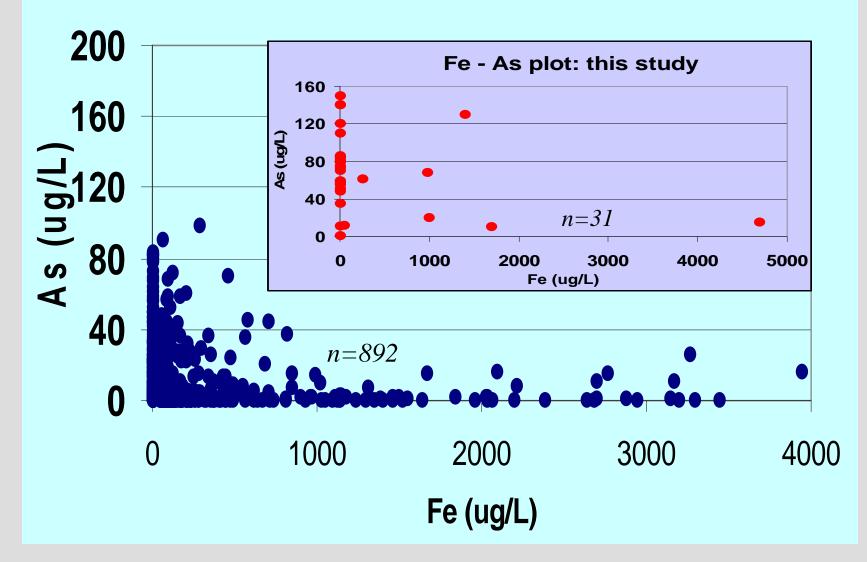
ORP - As(III): Inner Piedmont



Linear Regressions

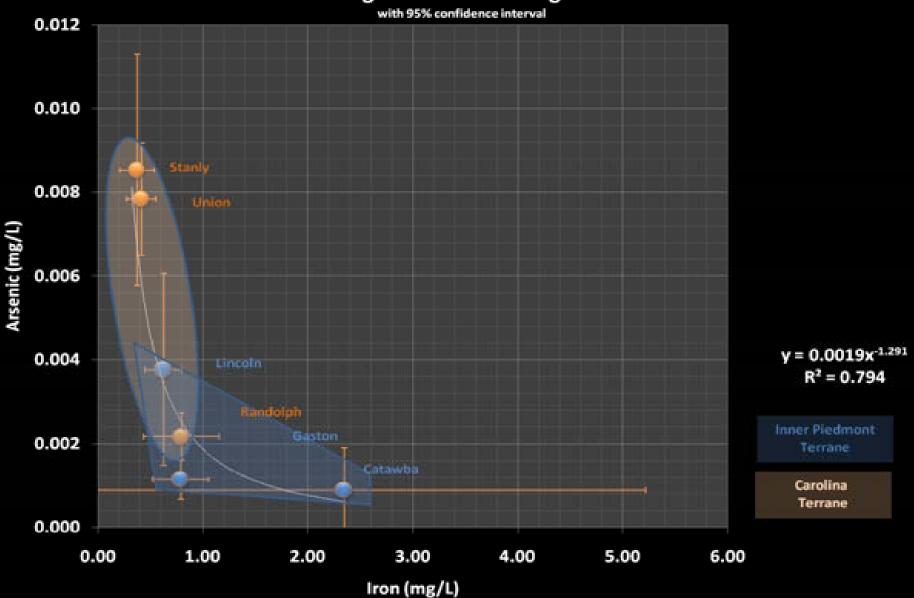


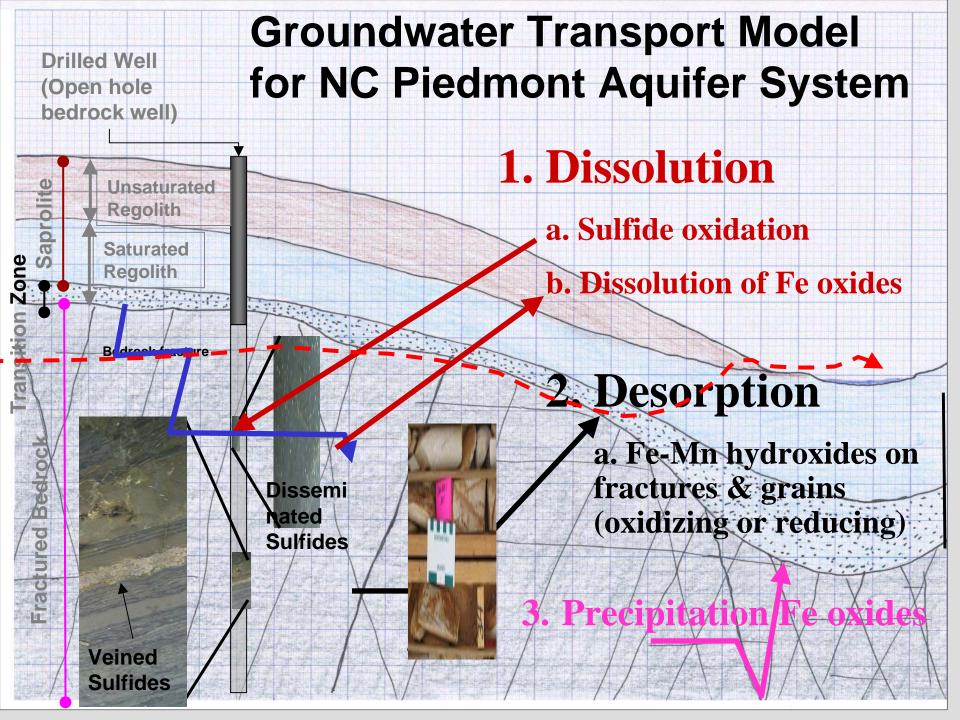
Iron - Arsenic plot: NCDHHS data



Regional Fe – As Relation







SUMMARY

- There exists a relationship between ORP, pH, Fe and As concentrations.
- Higher Fe levels present naturally in the IP groundwater appear to regulate As levels through adsorption or precipitation onto Fe hydroxides.
- Reductive dissolution of As or desorption of As from Fe oxyhydroxides appear to control As levels in the CT.

SUMMARY

- This study:
 - Provides new data on As concentration and speciation in private wells.
 - Developed gross relationships between redox parameters and As levels in the region.
 - Supports the hypothetical model on groundwater
 As transport in the NC Piedmont.
- Further investigations are needed to:
 - Test and verify the groundwater transport model